



Electrical Energy Storage using only Lunar Materials

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Permanently-occupied Lunar Base



■ POWER

- Must persist through lunar night (0354 hours)
- Must be highly-reliable, or redundant

■ Base-load options:

- Nuclear reactor
- Beamed power from orbiting SPS

■ Storage options:

- Batteries
- Hydrogen

Brief History of Batteries

- Count Alessandro Giuseppe Antonio Anastasio Volta
 - Invented the “voltaic pile” in 1800.
 - The term “**battery**” was coined by Benjamin Franklin in 1748
 - Described an array of charged glass plates
- Gaston Plante: **rechargeable** battery
 - Same lead-acid battery used in cars today – in 1859
- Thomas Edison: **alkaline** storage battery
 - The top selling battery today! – invented in 1901
- A “**battery**” is a collection of electrochemical “**cells**” wired together to boost power.
 - Cells inside a 9V battery



ISRU Battery Fabrication

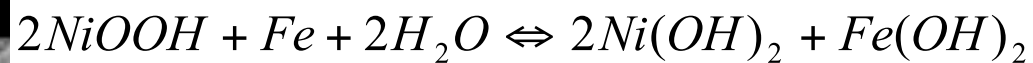
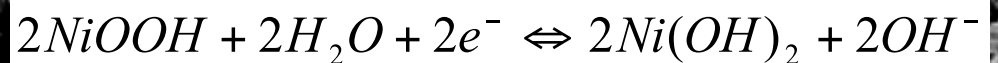
- Edison Cell

- Sold until '72



- Electrodes:

- Iron anode
- Nickel cathode



- Electrolyte:

- Potassium hydroxide in water

- Casing

- Cast basalt, or polymer-lined container

Electricity Storage Requirements

- Consider 2-person base
 - Average 1.2 kW
 - 100 sq. meters
- Equatorial location
- Horticulture:
 - 20 sq. meters per person
 - 100 W per sq. meter
- Inside 23 °C, outside 0-233 °C
 - With burial, outside temp is also 23 °C!



Fe and Ni – Method A

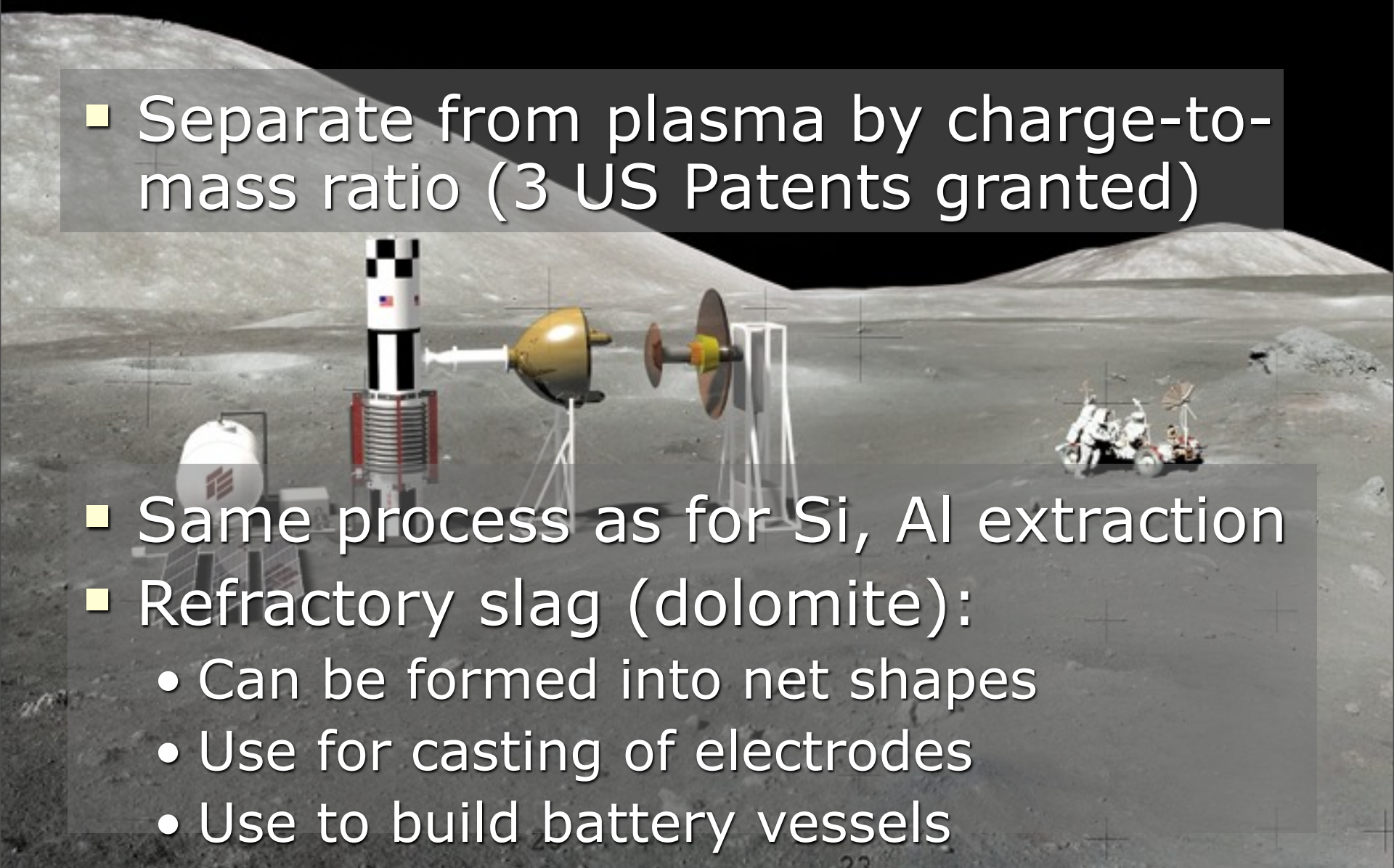
- Magnetically harvest fines
- Crack off silicates
 - Centrifugal grinder
- Mond process selects out Ni
 - At 60 °C
$$\text{Ni}(s) + 4\text{CO}(g) \rightleftharpoons \text{Ni}(\text{CO})_4(g)$$
- For further Fe refinement, use Mond process at 180 °C
- Precipitate metals at different T



Fe and Ni – Method B

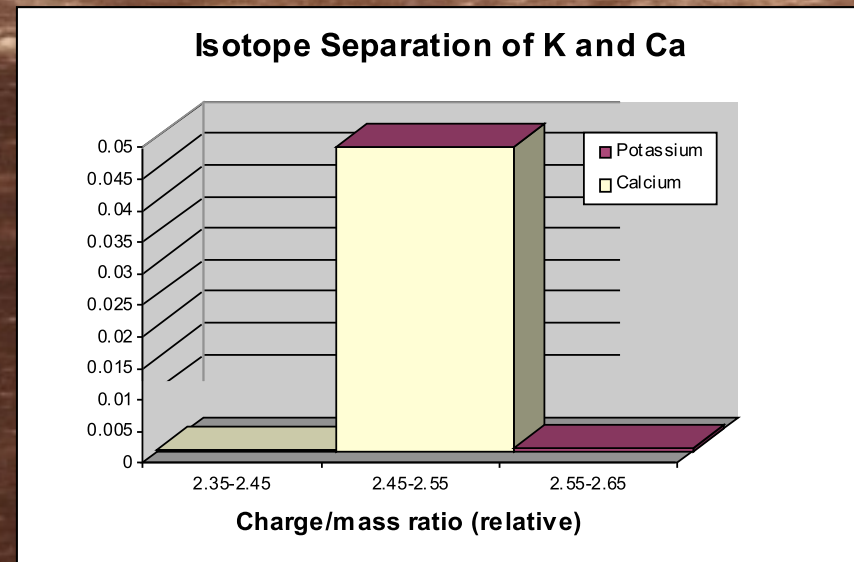
- Separate from plasma by charge-to-mass ratio (3 US Patents granted)

- Same process as for Si, Al extraction
- Refractory slag (dolomite):
 - Can be formed into net shapes
 - Use for casting of electrodes
 - Use to build battery vessels



Potassium Extraction

- Vacuum roasting between 900 and 1200 °C releases 30% of K.
- K is in lunar atmosphere, so we know it is volatile
- Alternate:
 - Isotope extraction
 - Separate by solubility
 - KOH: 110 g/100ml
 - CaOH: 0.17 g/100ml



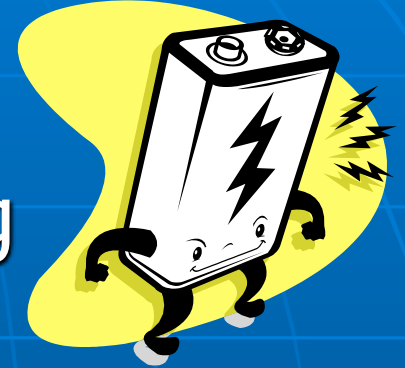
Water and Casing



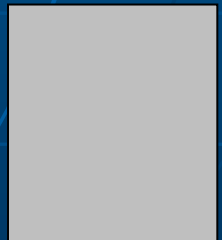
- Water abundant at poles only
 - Subsurface temperatures are low
 - Assume 85% insolation
 - With 2 meter masts, can get to 90%
- Vessels from cast basalt, dolomite, or polymer-lined vessels
 - Need a sealed lid
 - Assume 1 m³ H₂O each



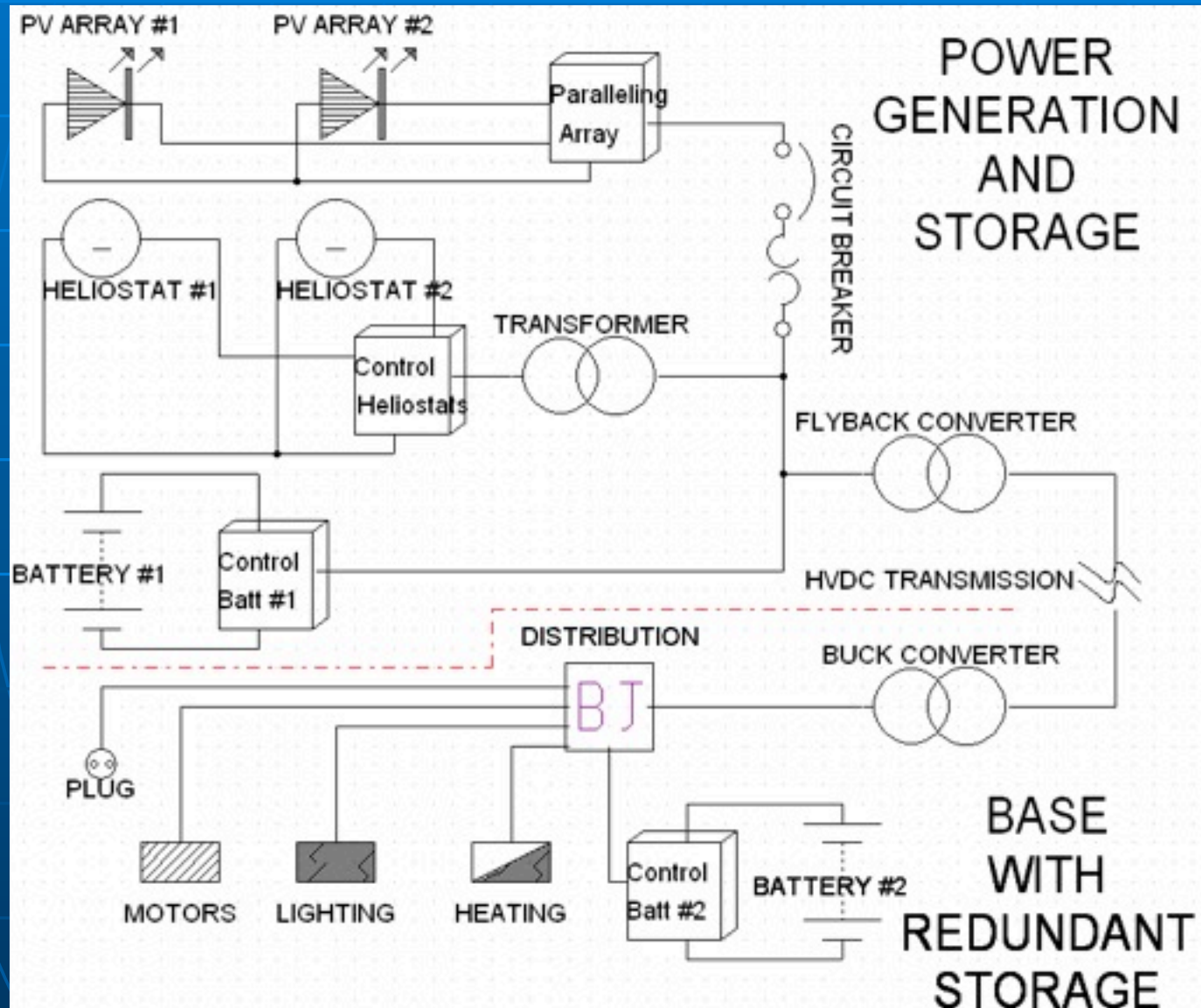
Battery Specifications



- Specific power = 0.04 kWh/kg
- Leakage rate = 30%/month
- Deep cycling limit = 65%
- Electrolyte specific density = 1.4
- Assuming 24 V = 21 cells/battery
- Electrode mass = 0.5 MT/battery
- Heat loss by radiation = 4.6 W/m²



Distribution Schematic



Results – 2-man base 85% sun

Energy Storage System		Base Needs	
electrolyte	1400 kg	Baseload	1.2 kW
Electrodes	500 kg	Energy needed	63.72 kWh
specific power	0.04 kWh/kg	Width	10 m
Storage	76 kWh/battery	Breadth	10 m
discharge	1% per day	height	3 m
T-background	40 K	Floorspace	100 m ²
T-inside	163 K	Volume	300 m ³
surface area	6 m ²	Surface area	320 m ²
heat loss	-0.04 kW/m ²	T-external	40 K
nightspan	53.1 hours	heat loss rate	12.76 kW
radiative loss	-12.71 kWh	heat loss	677.68 kWh
self-discharge	-1.68 kWh	plant needs	100 W/m ²
deep discharge	0.65 fraction	plant floor	50 m ²
Net storage	40.05 kWh/battery	duty cycle	0.5 half time
Batteries	6	Plant power	2.5 kW
Redundancy	12	Plant energy	132.75 kWh
		Total energy	209.23 kWh

Logistics



- Water harvesting assumed
- Iron extraction:
 - 16 batteries/year by isotope separation
 - *Also get Si, Al, K, plus dolomite slag*
 - 55 batteries/year by magnetic extraction
 - *Requires CO for carbonyls*
- Factory launch masses:
 - 1.3 MT for isotope separator
 - 0.4 MT for magnet harvester



CONCLUSIONS

- Electric energy storage is possible using only lunar materials, plus an earth-launched ISRU factory.
- In one year, we can serve up to 2-3 continuously-occupied, polar-located 2-person lunar bases.